

## **Glove Selection Guide**

The following Glove Selection & Usage Chart provides advantages and disadvantages for specific glove types. This guide was prepared for laboratory researchers but is helpful for all people working with hazardous materials.

**Always Read the Safety Data Sheets (SDSs) for each chemical involved.**

### **Glove Selection & Usage Chart**


What to do	How to do it
<b>Identify the hazards of the material(s) you'll be working with</b>	<p>Base selection of glove type and material on the type of exposure and the nature of the hazard. Some chemicals can easily penetrate gloves that work well for other chemicals.</p> <p>Consider these factors:</p> <ul style="list-style-type: none"> <li>• Chemical types</li> <li>• pH</li> <li>• Toxicity</li> <li>• Temperature extremes, cryogenic properties</li> <li>• Physical hazards (sharps, piercing objects)</li> <li>• Infectious potential of biological hazards</li> </ul>
<b>Determine if you will have incidental or extended contact with the hazardous materials</b>	<p>A. <b>Incidental Contact</b> includes these situations:</p> <ul style="list-style-type: none"> <li>• Accidental spill or splashes</li> <li>• Accidental overspray from a dispensing device</li> <li>• Handling infectious agents that require barrier protections</li> <li>• To prevent contamination of materials during handling</li> </ul> <p>B. <b>Extended Contact</b> includes these situations:</p> <ul style="list-style-type: none"> <li>• Handling highly contaminated materials</li> <li>• Submerging hands in a chemical or other hazardous substance</li> <li>• Need for physical protection from temperature extremes or sharp/piercing objects</li> </ul> <p>❖ <b>If you have incidental contact , go to Step 3</b></p> <p>❖ <b>If you have extended contact, go to Step 4</b></p>
<b>For incidental contact follow these selection guidelines</b>	<ol style="list-style-type: none"> <li>1. Type of glove: disposable, surgical-type gloves are appropriated for incidental contact.</li> <li>2. Nitrile gloves are preferred over latex because of their chemical resistance, their tendency to visibly rip when punctured and to prevent possible latex allergies.</li> </ol>






What to do	How to do it
	<p>3. Disposable gloves usage:</p> <ul style="list-style-type: none"> <li>• Check for rips or punctures before use</li> <li>• Remove and replace gloves immediately with new ones when a chemical spills or splashes on them</li> <li>• Never wash or reuse disposable gloves</li> <li>• Always remove gloves before touching objects such as door knobs, phones or elevator buttons</li> </ul>
<p><b>For extended contact</b> follow these guidelines</p>	<ol style="list-style-type: none"> <li>1. Type of glove: More substantial gloves are required for extended use.</li> <li>2. Norfoil gloves are recommended for highly toxic materials and materials that are absorbed through the skin.</li> <li>3. <a href="#">See Glove Comparison Chart</a> for advantages &amp; disadvantages if a commonly used gloves is used for extended contact.</li> <li>4. Reusable glove usage: <ul style="list-style-type: none"> <li>• Many gloves intended for extended contact are reusable</li> </ul> </li> </ol> <p>Check the gloves for:</p> <ul style="list-style-type: none"> <li>• Rips or punctures before and after each use</li> <li>• Prior contamination</li> <li>• Signs of degradation (change in color or texture)</li> <li>• Replace gloves as soon as signs of degradation appear</li> <li>• Wash after removal and air dry.</li> <li>• Consider wearing inner pair of gloves for extra protection</li> </ul>
<p><b>Dispose of used and damaged gloves</b> according to whether or not they're contaminated with a hazardous material</p>	<p style="text-align: center;"><b><i>ALWAYS wash your hands after removing gloves.</i></b></p>



## **Glove Comparison Chart**

Consult this chart for an overview of commonly used glove types for laboratory use and their general advantages and disadvantages.

NOTE: Pictures are examples and glove appearance and color will vary.

Glove Material	Intended Use	Advantages & Disadvantages	Example Photos
Latex (natural rubber) gloves	Incidental Contact	<ul style="list-style-type: none"> <li>• Good for biological &amp; water-based materials</li> <li>• Poor for organic solvents</li> <li>• Little chemical protection</li> <li>• Hard to detect puncture holes</li> <li>• Can cause or trigger latex allergies</li> </ul>	
Nitrile gloves	Incidental contact (disposable exam glove)  Extended contact (heavier, reusable glove)	<ul style="list-style-type: none"> <li>• Excellent general use glove. Good for solvents, oils, greases and some acids and bases</li> <li>• Clear indication of tears and breaks</li> <li>• Good alternative for those with <i>latex allergies</i></li> </ul>	
Butyl rubber gloves	Extended contact	<ul style="list-style-type: none"> <li>• Good for ketones and esters</li> <li>• Poor for gasoline and aliphatic, aromatic and halogenated hydrocarbons</li> </ul>	
Neoprene gloves	Extended contact	<ul style="list-style-type: none"> <li>• Good For acids, bases, alcohols, fuels, peroxides, hydrocarbons and phenols</li> <li>• Poor for halogenated &amp; aromatic hydrocarbons</li> <li>• Good for most hazardous chemicals</li> </ul>	

Glove Material	Intended Use	Advantages & Disadvantages	Example Photos
Norfoil	Extended Contact	<ul style="list-style-type: none"> <li>• Good for most hazardous chemicals</li> <li>• Poor fit. Dexterity can be partially regained by using a heavier weight nitrile glove over the Norfoil/Silver Shield glove</li> </ul>	
Viton	Extended contact	<ul style="list-style-type: none"> <li>• Good for chlorinated &amp; aromatic solvents</li> <li>• Good resistance to cut and abrasions</li> <li>• Poor for ketones</li> <li>• Expensive</li> </ul>	
Polyvinyl chloride (PVC) gloves	Specific use	<ul style="list-style-type: none"> <li>• Good for acids, bases, oils, fats, peroxides and amines</li> <li>• Good resistance to abrasions</li> <li>• Poor for most organic solvents</li> </ul>	
Polyvinyl alcohol (PVA) gloves	Specific use	<ul style="list-style-type: none"> <li>• Good for aromatic &amp; chlorinated solvents</li> <li>• Poor for water-based solutions</li> </ul>	
Stainless steel Kevlar Leather	Specific use	<ul style="list-style-type: none"> <li>• Cut-resistant gloves</li> <li>• Sleeves are also available to provide protection to wrists &amp; forearms</li> <li>• If potential for biological or chemical contamination, wear appropriate disposable gloves on top of your cut-resistant gloves and discard after use</li> </ul>	

Glove Material	Intended Use	Advantages & Disadvantages	Example Photos
<p>Cryogenic Resistant Materials gloves</p> <p>Leather</p>	<p>Specific use</p>	<ul style="list-style-type: none"> <li>• For use with cryogenic materials</li> <li>• Designed to prevent frostbite.</li> </ul> <p><b>NOTE: Never dip gloves directly into liquid nitrogen</b></p>	
<p>Nomex</p>	<p>Specific use</p>	<ul style="list-style-type: none"> <li>• For use with pyrophoric materials</li> <li>• Consider wearing a flame-resistant glove such as Nomex “flight” gloves with a thin nitrile exam glove underneath</li> </ul>	

## Glove Type and Chemical Use

**\*\*Always check the product SDS to verify that the appropriate glove has been correctly selected for the job\*\***

<b>*Limited services</b>	<b>VG=Very Good</b>	<b>G=Good</b>	<b>F=Fair</b>	<b>P=Poor (not recommended)</b>
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Chemical	Neoprene	Natural Latex or Rubber	Butyl	Nitrile
<i>*Acetaldehyde</i>	VG	G	VG	G
<i>Acetic acid</i>	VG	VG	VG	VG
<i>*Acetone</i>	G	VG	BG	P
<i>Ammonium Hydroxide</i>	VG	VG	VG	VG
<i>*Amyl Acetate</i>	F	P	F	P
<i>Aniline</i>	G	F	F	P
<i>*Benzaldehyde</i>	F	F	G	G
<i>*Benzene</i>	F	F	F	P
<i>Butyl acetate</i>	G	F	F	P
<i>Butyl alcohol</i>	VG	VG	VG	VG
<i>Carbon Disulfide</i>	F	F	F	F
<i>*Carbon Tetrachloride</i>	F	P	P	G
<i>Castor oil</i>	F	P	F	VG
<i>*Chlorobenzene</i>	F	P	F	P
<i>*Chloroform</i>	G	P	P	P
<i>Chloronaphthalene</i>	F	P	F	F
<i>Chromic Acid (50%)</i>	F	P	F	F
<i>Citric Acid (10%)</i>	VG	VG	VG	VG
<i>Cyclohexanol</i>	G	F	G	VG
<i>*Dibutyl Phthalate</i>	G	P	G	G
<i>Diesel Fuel</i>	G	P	P	VG
<i>Diisobutyl Ketone</i>	P	F	G	P
<i>Dimethylformamide</i>	F	F	G	G
<i>Dioctyl Phthalate</i>	G	P	F	VG
<i>Dioxane</i>	VG	G	G	G
<i>Epoxy resins, dry</i>	VG	VG	VG	VG
<i>*Ethyl acetate</i>	G	F	G	F
<i>Ethyl Alcohol</i>	VG	VG	VG	VG
<i>Ethyl Ether</i>	VG	G	VG	G
<i>*Ethylene dichloride</i>	F	P	F	P
<i>Ethylene Glycol</i>	VG	VG	VG	VG
<i>Formaldehyde</i>	VG	VG	VG	VG
<i>Formic Acid</i>	VG	VG	VG	VG
<i>Freon 11</i>	G	P	F	G
<i>Freon 12</i>	G	P	F	G
<i>Freon 21</i>	G	P	F	G
<i>Freon 22</i>	G	P	F	G

<i>Chemical</i>	<i>Neoprene</i>	<i>Natural Latex or Rubber</i>	<i>Butyl</i>	<i>Nitrile</i>
<i>*Furfural</i>	<i>G</i>	<i>G</i>	<i>G</i>	<i>G</i>
<i>Gasoline, leaded</i>	<i>G</i>	<i>P</i>	<i>F</i>	<i>VG</i>
<i>Gasoline, unleaded</i>	<i>G</i>	<i>P</i>	<i>F</i>	<i>VG</i>
<i>Glycerin</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Hexane</i>	<i>F</i>	<i>P</i>	<i>P</i>	<i>G</i>
<i>Hydrochloric Acid</i>	<i>VG</i>	<i>G</i>	<i>G</i>	<i>G</i>
<i>Hydrofluoric Acid (48%)</i>	<i>VG</i>	<i>G</i>	<i>G</i>	<i>G</i>
<i>Hydrogen Peroxide (30%)</i>	<i>G</i>	<i>G</i>	<i>G</i>	<i>G</i>
<i>Hydroquinone</i>	<i>G</i>	<i>G</i>	<i>G</i>	<i>F</i>
<i>Isooctane</i>	<i>F</i>	<i>P</i>	<i>P</i>	<i>VG</i>
<i>Isopropyl alcohol</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Kerosene</i>	<i>VG</i>	<i>F</i>	<i>F</i>	<i>VG</i>
<i>Ketones</i>	<i>G</i>	<i>VG</i>	<i>VG</i>	<i>P</i>
<i>Lacquer Thinner</i>	<i>G</i>	<i>F</i>	<i>F</i>	<i>P</i>
<i>Lactic Acid (85%)</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Lauric Acid (36%)</i>	<i>VG</i>	<i>F</i>	<i>VG</i>	<i>VG</i>
<i>Lineoleic Acid</i>	<i>VG</i>	<i>P</i>	<i>F</i>	<i>G</i>
<i>Linseed Oil</i>	<i>VG</i>	<i>P</i>	<i>F</i>	<i>VG</i>
<i>Maleic Acid</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Methyl Alcohol</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Methylamine</i>	<i>F</i>	<i>F</i>	<i>G</i>	<i>G</i>
<i>Methyl Bromide</i>	<i>G</i>	<i>F</i>	<i>G</i>	<i>F</i>
<i>*Methyl Chloride</i>	<i>P</i>	<i>P</i>	<i>P</i>	<i>P</i>
<i>*Methyl Ethyl Ketone</i>	<i>G</i>	<i>G</i>	<i>VG</i>	<i>P</i>
<i>*Methyl Isobutyl Ketone</i>	<i>F</i>	<i>F</i>	<i>VG</i>	<i>P</i>
<i>Methyl methacrylate</i>	<i>G</i>	<i>G</i>	<i>VG</i>	<i>F</i>
<i>Monoethanolamine</i>	<i>VG</i>	<i>G</i>	<i>VG</i>	<i>VG</i>
<i>Morpholine</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>G</i>
<i>Naphthalene</i>	<i>G</i>	<i>F</i>	<i>F</i>	<i>G</i>
<i>Naphtha, aliphatic</i>	<i>VG</i>	<i>F</i>	<i>F</i>	<i>VG</i>
<i>Naphtha, aromatics</i>	<i>G</i>	<i>P</i>	<i>P</i>	<i>G</i>
<i>*Nitric Acid</i>	<i>G</i>	<i>F</i>	<i>F</i>	<i>F</i>
<i>Nitromethane (95%)</i>	<i>F</i>	<i>P</i>	<i>F</i>	<i>F</i>
<i>Nitropropane (95%)</i>	<i>F</i>	<i>P</i>	<i>F</i>	<i>F</i>
<i>Octyl Alcohol</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Oleic Acid</i>	<i>VG</i>	<i>F</i>	<i>G</i>	<i>VG</i>
<i>Oxalic Acid</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Palmitic Acid</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Perchloric Acid (60%)</i>	<i>VG</i>	<i>F</i>	<i>G</i>	<i>G</i>
<i>Perchloroethylene</i>	<i>F</i>	<i>P</i>	<i>P</i>	<i>G</i>
<i>Petroleum distillates (Naphtha)</i>	<i>G</i>	<i>P</i>	<i>P</i>	<i>VG</i>
<i>Phenol</i>	<i>VG</i>	<i>F</i>	<i>G</i>	<i>F</i>
<i>Phosphoric Acid</i>	<i>VG</i>	<i>G</i>	<i>VG</i>	<i>VG</i>
<i>Potassium Hydroxide</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Propyl Acetate</i>	<i>G</i>	<i>F</i>	<i>G</i>	<i>F</i>
<i>Propyl Alcohol</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Propyl Alcohol (iso)</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Sodium Hydroxide</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Styrene</i>	<i>P</i>	<i>P</i>	<i>P</i>	<i>F</i>

<i>Chemical</i>	<i>Neoprene</i>	<i>Natural Latex or Rubber</i>	<i>Butyl</i>	<i>Nitrile</i>
<i>Styrene (100%)</i>	<i>P</i>	<i>P</i>	<i>P</i>	<i>F</i>
<i>Sulfuric Acid</i>	<i>G</i>	<i>G</i>	<i>G</i>	<i>G</i>
<i>Tannic Acid (65%)</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>	<i>VG</i>
<i>Tetrahydrofuran</i>	<i>P</i>	<i>F</i>	<i>F</i>	<i>F</i>
<i>*Toluene</i>	<i>F</i>	<i>P</i>	<i>P</i>	<i>F</i>
<i>Toluene diisocyanate</i>	<i>F</i>	<i>G</i>	<i>G</i>	<i>F</i>
<i>*Trichloroethylene</i>	<i>F</i>	<i>F</i>	<i>P</i>	<i>G</i>
<i>Triethanolamine</i>	<i>VG</i>	<i>G</i>	<i>G</i>	<i>F</i>
<i>Tung Oil</i>	<i>VG</i>	<i>P</i>	<i>F</i>	<i>VG</i>
<i>Turpentine</i>	<i>G</i>	<i>F</i>	<i>F</i>	<i>VG</i>
<i>*Xylene</i>	<i>P</i>	<i>P</i>	<i>P</i>	<i>F</i>